

SHORT
COMMUNICATIONS

New Route of Reaction between Acyl Isothiocyanates and Cyclic β -Diketones

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The reaction of acyl isothiocyanates with β -dicarbonyl compounds is thoroughly studied nowadays [1–10]. In a wide range of reaction conditions the process proceeds as C-thiocarbamylation of the dicarbonyl component. For instance, C-thiocarbamoylated derivatives of 1,3-cyclohexanedione [7] or dimedone [8, 9] formed in good yields at boiling reagents in dioxane or acetone.

We found reaction conditions where the character of the process crucially changed: the acyl isothiocyanates operated not as thiocarbamoylating but as acylating agents. It was demonstrated that reaction of benzoyl isothiocyanates **Ia–Id** with cyclic diketones **IIa, IIb** in the presence of equimolar amount of KOH at room temperature furnished O-acylation products **IIIa–IIIf**.

We assume as more probable a direct O-acylation of β -diketones of **II** type with compounds **I** whose isothiocyanate group under the reaction conditions

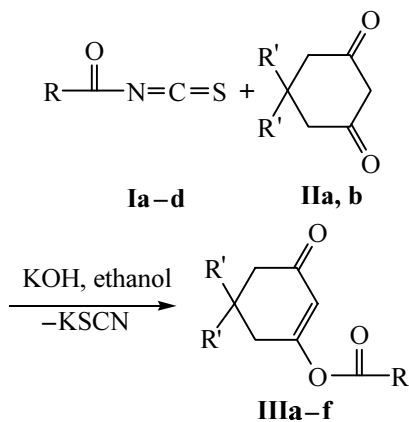
(alcohol solution of KOH) acts as pseudohaloid. We believe that conversion of C-carbamoylated products into O-acylated derivatives under the action of alkali is less probable. The latter is confirmed by formation of derivatives of S,N-ketene acetal type in reaction between benzoyl isothiocyanate with β -dicarbonyl compounds in DMF in the presence of KOH [3].

Compounds **IIIa–IIIf** are viscous oily fluids. Their structure was confirmed by IR spectra containing characteristic absorption bands of keto ($1665\text{--}1675\text{ cm}^{-1}$) and ester ($1725\text{--}1740\text{ cm}^{-1}$) carbonyl groups, and by ^1H NMR spectra where appeared a signal from proton $\text{CH}=\text{C}$ at $5.94\text{--}6.06\text{ ppm}$

3-Acyloxy-2-cyclohexen-1-ones (IIIa–IIIf). To a solution of 0.006 mol of β -diketone **IIa, IIb** and 0.337 g (0.006 mol) of KOH in 30 ml of ethanol was added at stirring in one portion 0.006 mol of benzoyl isothiocyanate **Ia–d**. After stirring for 15 min the reaction mixture was poured into water (50 ml), the organic products were extracted into dichloromethane ($2 \times 25\text{ ml}$), the extract was dried on Na_2SO_4 , and the solvent was evaporated. Then the residue was extracted with boiling hexane (100 ml), and on evaporating the solvent the substance was subjected to chromatography (eluent ethylacetate–hexane, 1:2).

3-Benzoyloxy-2-cyclohexen-1-one (IIIa). Yield 76%, R_f 0.82. IR spectrum, cm^{-1} : 1675, 1730 (C=O). ^1H NMR spectrum, δ , ppm: 2.07–2.11m (2H, H^5), 2.38–2.41m (2H, H^6), 2.70–74m (2H, H^4), 5.91s (1H, $\text{CH}=\text{C}$), 7.59–7.73m (3H, $\text{H}^{3'}$, $\text{H}^{4'}$, $\text{H}^{5'}$), 8.09d (2H, $\text{H}^{2'}$, $\text{H}^{6'}$, J 7.8 Hz). Found, %: C 71.83; H 5.72. $\text{C}_{13}\text{H}_{12}\text{O}_3$. Calculated, %: C 72.21; H 5.59.

3-(4-Chlorobenzoyloxy)-2-cyclohexen-1-one (IIIb). Yield 61%, R_f 0.60. IR spectrum, cm^{-1} : 1675, 1735



I, R = Ph (**a**), 4-ClC₆H₄ (**b**), 4-BrC₆H₄ (**c**), 4-MeOC₆H₄ (**d**); **II**, R' = H (**a**), Me (**b**); **III**, R' = H, R = Ph (**a**), 4-ClC₆H₄ (**b**), R' = Me, R = Ph (**c**), 4-ClC₆H₄ (**d**), 4-BrC₆H₄ (**e**), 4-MeOC₆H₄ (**f**).

(C=O). ^1H NMR spectrum, δ , ppm: 2.10–2.14 m (2H, H^5), 2.46 t (2H, H^6 , J 5.8 Hz), 2.67 t (2H, H^4 , J 5.8 Hz), 6.04 c (1H, CH=), 7.47 d (2H, H^3 , H^5 , J 8.0 Hz), 8.02 d (2H, H^2 , H^6 , J 8.0 Hz). Found, %: C 62.59; H 4.30. $\text{C}_{13}\text{H}_{11}\text{ClO}_3$. Calculated, %: C 62.29; H 4.42.

3-Benzoyloxy-5,5-dimethyl-2-cyclohexen-1-one (IIIc). Yield 72%. R_f 0.71. IR spectrum, cm^{-1} : 1670, 1725 (C=O). ^1H NMR spectrum, δ , ppm: 1.16 s (6H, 2Me), 2.33 s (2H, H^6), 2.56 s (2H, H^4), 6.06 s (1H, CH=), 7.50–7.53 m (2H, H^3 , H^5), 7.65 t (1H, H^4 , J 7.8 Hz), 8.08 d (2H, H^2 , H^6 , J 7.7 Hz). Found, %: C 74.07; H 6.77. $\text{C}_{15}\text{H}_{16}\text{O}_3$. Calculated, %: C 73.75; H 6.60.

3-(4-Chlorobenzoyloxy)-5,5-dimethylcyclohexen-1-one (IIIId). Yield 63%, R_f 0.74. IR spectrum, cm^{-1} : 1665, 1730 (C=O). ^1H NMR spectrum, δ , ppm: 1.16 s (6H, 2Me), 2.33 s (2H, 4- CH_2), 2.55 s (2H, H^4), 6.05 s (1H, CH=), 7.47 d (2H, H^3 , H^5 , J 8.0 Hz), 8.02 d (2H, H^2 , H^6 , J 8.0 Hz) Found, %: C 64.52; H 5.60. $\text{C}_{15}\text{H}_{15}\text{ClO}_3$. Calculated, %: C 64.64; H 5.42.

3-(4-Bromobenzoyloxy)-5,5-dimethyl-2-cyclohexen-1-one (IIIe). Yield 60%, R_f 0.8. IR spectrum, cm^{-1} : 1670, 1725 (C=O). ^1H NMR spectrum, δ , ppm: 1.16 s (6H, 2Me), 2.34 s (2H, H^6), 2.55 s (2H, H^4), 6.05 s (1H, CH=), 7.64 d (2H, H^3 , H^5 , J 7.8 Hz), 7.93 d (2H, H^2 , H^6 , J 7.8 Hz) Found, %: C 55.98; H 4.86. $\text{C}_{15}\text{H}_{15}\text{BrO}_3$. Calculated, %: C 55.75; H 4.68.

3-(4-Methoxybenzoyloxy)-5,5-dimethyl-2-cyclohexen-1-one (IIIIf). Yield 65%, R_f 0.70. IR spectrum, cm^{-1} : 1665, 1740 (C=O). ^1H NMR spectrum,

δ , ppm: 1.15 s (6H, 2Me), 2.33 s (2H, H^6), 2.53 s (2H, H^4), 6.04 s (1H, CH=), 6.97 d (2H, H^3 , H^5 , J 8.0 Hz), 8.03 d (2H, H^2 , H^6 , J 8.0 Hz) Found, %: C 69.73; H 6.47. $\text{C}_{16}\text{H}_{18}\text{O}_4$. Calculated, %: C 70.06; H 6.61.

IR spectra were recorded on spectrophotometer UR-20 from solutions in CH_2Cl_2 . ^1H NMR spectra were registered on spectrometer Varian-Gemini (300.0 MHz) in CDCl_3 , internal reference TMS.

REFERENCES

1. Dehne, H. and Krey, H., *J. Pract. Chem.*, 1982, vol. 324, p. 915.
2. Mohareb, R.M., Habashi, A., Ibrahim, N.S., and Sherif, S.M. *Synthesis*, 1987, p. 228.
3. Mohareb, R.F., Shams, H.H., and Azis, S.I., *J. Chem. Res.*, 1992, vol. 5, p. 1132.
4. Mohareb, R.F., Azis, S.I., Abdel-Sayed, N.I., and El-Ab-lack, F.Z., *Collect. Cheh. Chem. Commun.*, 1993, vol. 58, p. 947.
5. Assy, M.G. and Moustafa, H.Y., *Phosphorus, Sulfur, Silicon*, 1995, vol. 105, p. 213.
6. Hetaba, A.A., Assy, M.G., and Fikry, R.M., *Ind. J. Chem. B*, 1996, vol. 35, p. 144.
7. Assy, M.G., *Phosphorus, Sulfur, Silicon*, 1996, vol. 108, p. 15.
8. Assy, M.G. and Motti, F.M., *Egypt. J. Chem.*, 1996, vol. 39, 581.
9. Assy, M.G. and Hetaba, A.A., *J. Ind. Chem. Soc.*, 1997, vol. 74, p. 239.
10. Amine M.S. and Arief M.M.H., *Ind. J. Chem. B*, 1998, vol. 37, p. 135.